

ISOLATION AND STRUCTURE OF AFLATOXINS M<sub>1</sub> AND M<sub>2</sub>

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A toxic factor is excreted in the milk of cows fed aflatoxin containing groundnut meal<sup>1</sup>. de Jongh *et al.*<sup>2</sup> showed that the toxic factor is a blue-violet fluorescent compound which had an  $R_F$  value well below that of aflatoxin B<sub>1</sub>. They also found that the lactating rat was able to convert aflatoxin B<sub>1</sub> into this 'milk toxin', while Butler and Clifford<sup>3</sup> have shown the presence of this component in the livers of rats given aflatoxin B<sub>1</sub>. Allcroft *et al.*<sup>4</sup> presented chromatographic evidence that a compound excreted in the urine of sheep fed mixed aflatoxins was probably identical to the 'milk toxin' and suggested giving the compound the trivial name 'aflatoxin M'.

We have repeated the experiments of Allcroft *et al.* Two adult sheep (total weight, 67 kg) were each given intraperitoneally a dose (1 mg/kg) of mixed aflatoxins (B<sub>1</sub> 73%, B<sub>2</sub> 24%, G<sub>1</sub> 2% and G<sub>2</sub> 1%) dissolved in arachis oil (2 mg/ml). The urine collected during the 48 hr. after dosing was extracted continuously with chloroform for 36 hr. and the extract chromatographed on Merck's kieselgel. The material eluted with 2% methanol in chloroform contained aflatoxin M as shown by thin-layer chromatography using the systems described<sup>4</sup> by Allcroft *et al.* A pure concentrate (0.7 mg) of aflatoxin M was

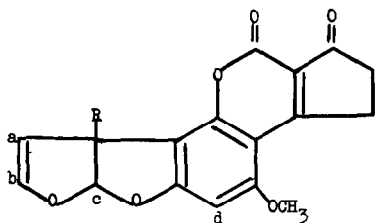
obtained by chromatography on Merck's kieselgel G chromatoplates using chloroform-methanol (97:3) as developing solvent. Chromatography of the concentrate on Whatman No. 1 filter paper impregnated with formamide-water (85:15) using ethyl acetate-benzene (9:1) as mobile phase gave two components; a blue-violet fluorescent substance ( $R_F$  0.34) and a violet fluorescent substance ( $R_F$  0.23) in the ratio of approximately 3:1. These components are designated aflatoxin  $M_1$  ( $R_F$  0.34) and aflatoxin  $M_2$  ( $R_F$  0.23).

de Iongh *et al.*<sup>2</sup> have presented chromatographic evidence that an extract of Aspergillus flavus culture on peanuts contained a component probably identical with aflatoxin M. We have extracted mouldy peanuts (7.5 kg) with hexane and then with acetone-water (86:14). The acetone extract contained aflatoxins  $B_1$ ,  $B_2$ ,  $G_1$ ,  $G_2$  as well as the unidentified blue-violet fluorescent component described by de Iongh *et al.* A concentrate of the latter substance was obtained by chromatography on formamide-impregnated cellulose powder (elution with chloroform) and then on Merck's kieselgel (elution with 2% methanol in chloroform). This concentrate (200 mg) was chromatographed on impregnated Whatman No. 3 M M filter paper using the solvent system described above. A blue-violet and a violet fluorescent band at  $R_F$  0.34 and 0.23 respectively were separately eluted (aqueous methanol), each yielding a pure crystalline compound. The compound with  $R_F$  0.34 (11 mg) was shown to be identical with the aflatoxin  $M_1$  extracted from urine. This was proved by paper and thin-layer chromatography, ultraviolet, infrared and mass spectra. Similarly, the compound with  $R_F$  0.23 (4 mg) was identical with aflatoxin  $M_2$ .

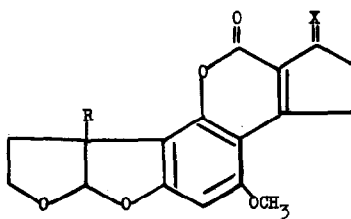
Aflatoxin  $M_1$  (from methanol) had m.p. 299° (decomp.),  $[\alpha]_D^{280}$  (c 0.1 in dimethylformamide)<sup>3</sup>,  $\lambda_{max}$  (ethanol) 226, 265 and 357 m $\mu$  (e 23,100,

<sup>3</sup>Aflatoxin  $B_1$   $[\alpha]_D^{480}$  (c 0.1 in dimethylformamide).

11,600 and 19,000 respectively) and was shown by microanalysis and mass spectrum to have the empirical formula  $C_{17}H_{12}O_7$ , one oxygen atom more than aflatoxin  $B_1$  (I, R = H)<sup>5</sup>. The close resemblance of the ultraviolet spectral data with that of aflatoxin  $B_1$  suggested that the main chromophoric system was similar. Its infrared spectrum (in nujol) resembled that of aflatoxin  $B_1$  and had carbonyl peaks at 1760 and 1690  $cm^{-1}$ . However, absorption at 3425  $cm^{-1}$  indicated that aflatoxin  $M_1$  contained a hydroxyl group. This was confirmed by reaction with acetic anhydride-pyridine (under conditions which left aflatoxin  $B_1$  unchanged) which gave an acetate, mol. weight 370 (mass spec.),  $\nu_{max}$ . (chloroform) 1760, 1740 (sh) and 1692  $cm^{-1}$ ,  $\lambda_{max}$ . (ethanol) 226, 265 and 357  $m\mu$ .



(I)



(II)

The above evidence led to the conclusion that aflatoxin  $M_1$  is hydroxy-aflatoxin  $B_1$  and it was formulated as (I, R = OH) on the basis of its nuclear magnetic resonance spectrum. The significant peaks in the spectrum (measured in deuteriodimethylsulphoxide) are as follows: methoxy group,  $\tau$  6.02; olefinic proton  $H_a$  -doublet at  $\tau$  4.36 ( $J = 3$  c./sec.); olefinic proton

$H_b$  - doublet at  $\tau$  3.17 ( $J = 3$  c./sec.);  $H_c$  and  $H_d$  singlets at  $\tau$  3.54 and 3.22. The fragmentation of aflatoxin  $M_1$  in the mass spectrometer is consistent with the formulation (I, R = OH) and, except for at peak  $m/e$  310 arising from the loss of a molecule of water from the parent molecule, is largely similar to that of aflatoxin  $B_1$ , but shifted 16 mass units higher.

Aflatoxin  $M_2$  (from methanol-chloroform) had m.p.  $293^\circ$  (decomp.),  $\nu_{\max.}$  (chloroform) 3350 (b), 1760 and 1690  $\text{cm}^{-1}$ ,  $\lambda_{\max.}$  (ethanol) 221, 264 and 357  $\mu$  ( $\epsilon$  20,000, 10,900 and 21,000 respectively), and mol. weight 330 (mass spec.). It was identical with dihydro-aflatoxin  $M_1$ , obtained by hydrogenation (10 min.) of aflatoxin  $M_1$  (Pd/C in acetic acid). The infrared spectrum of this compound lacked the vinyl ether bands at 3100, 1067 and 722  $\text{cm}^{-1}$  which are present in the spectrum of aflatoxin  $M_1$ . Aflatoxin  $M_2$  is, therefore, formulated as (II, R = OH, X = O).

Aflatoxin  $M_2$  is recovered unchanged on treatment with chromium trioxide in acetone, confirming the tertiary nature of the hydroxyl group. Prolonged hydrogenation of aflatoxin  $M_1$  (Pd/C in acetic acid) gave tetrahydro-desoxy-aflatoxin  $M_1$  (II, R = OH, X =  $H_2$ ), m.p.  $249-251^\circ$  (from benzene),  $\nu_{\max.}$  (chloroform) 3350 (b), and 1707  $\text{cm}^{-1}$ ,  $\lambda_{\max.}$  (ethanol), 254, 263 and 328  $\mu$  ( $\epsilon$  7,150, 8,000 and 13,000 respectively) and mol. weight 316 (mass spec.). Its nuclear magnetic resonance spectrum (in deuteriochloroform) showed peaks due to the acetal and aromatic protons (singlets at  $\tau$  3.65 and 3.79). Furthermore, the portion of the spectrum representing the six cyclopentene protons was identical in detail with the corresponding region in the spectrum of tetrahydro-desoxy-aflatoxin  $B_1$  (II, R = H, X =  $H_2$ ), m.p.  $248-249^\circ$  (lit.<sup>5</sup> m.p.  $249-250^\circ$ ).

Aflatoxin M in extracts of infected peanuts can be detected by

chromatography on Merck's kieselgel G chromatoplates employing chloroform-methanol (97:3) as the developing solvent. It can be quantitatively assayed by comparing the intensity of the fluorescent spot at  $R_F$  0.4 with that of standard aflatoxin M. The fluorescence of aflatoxins  $M_1$  and  $M_2$  is approximately three times the intensity of aflatoxin  $B_1$ .

The  $LD_{50}$  values of aflatoxin  $B_1$ ,  $M_1$  and  $M_2$  on day-old Pekin ducklings were determined simultaneously using Weil's<sup>6</sup> technique. The value for  $B_1$  was 12  $\mu$ g, for  $M_1$  was 16.6  $\mu$ g and for  $M_2$  was 62  $\mu$ g per day-old duckling.

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#### REFERENCES

1. R. Allcroft and R.B.A. Carnaghan; Vet. Rec. 75, 259 (1963).
2. H. de Iongh, R.O. Vles, and J.G. van Pelt; Nature 202, 466 (1964).
3. W.H. Butler and J.I. Clifford; Nature 206, 1045 (1965).
4. R. Allcroft, H. Rogers, G. Lewis, J. Nabney, and P.E. Best; Nature 209, 154 (1966).
5. T. Aseo, G. Bñchi, M.M. Abdel-Kader, S.B. Chang, E.L. Wick and G.N. Wogan; J. Amer. Chem. Soc. 87, 882 (1965).
6. C.S. Weil; Biometrics 8, 249 (1952).